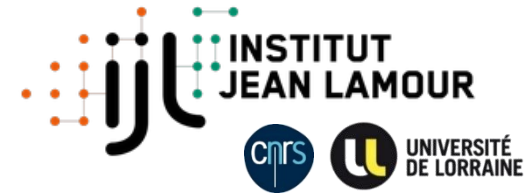




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Effect of V on the carbon distribution in carbide-free bainite during isothermal transformations

Irina Pushkareva¹, Juan Macchi², Babak Shalchi-Amirkhiz¹, Fateh Fazeli¹, Guillaume Geandier², Frédéric Danoix³, Julien Da Costa Teixeira, Sébastien Yves Pierre Allain², and Colin Scott¹

¹CanmetMATERIALS, Natural Resources Canada, Hamilton, Canada

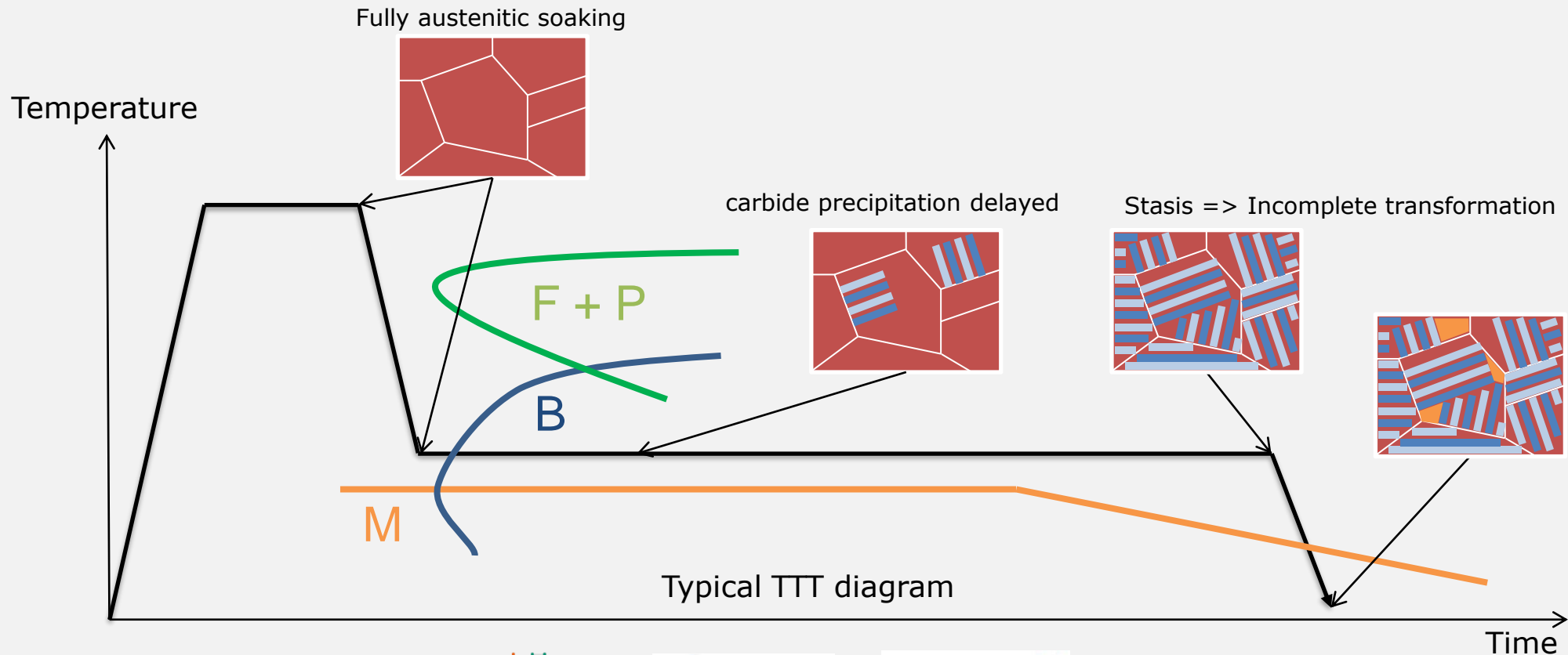
²Institut Jean Lamour, UMR CNRS-UL 7198, Nancy, France

³Normandie Université, UNIROUEN, INSA Rouen, CNRS, Groupe de Physique des Matériaux, Rouen France

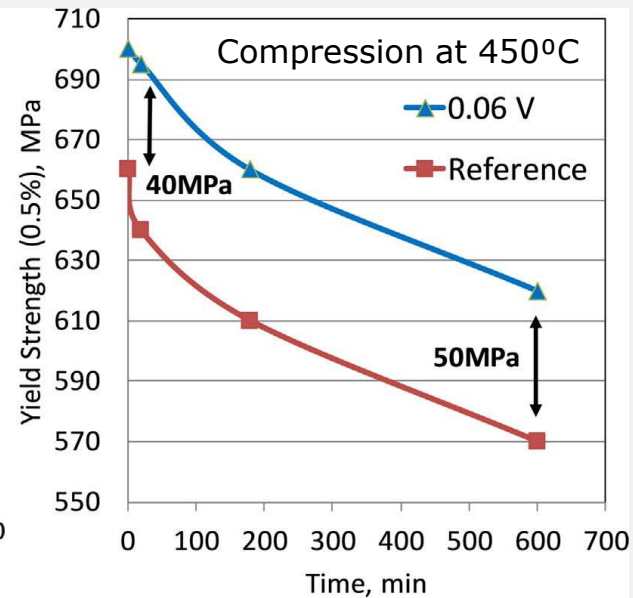
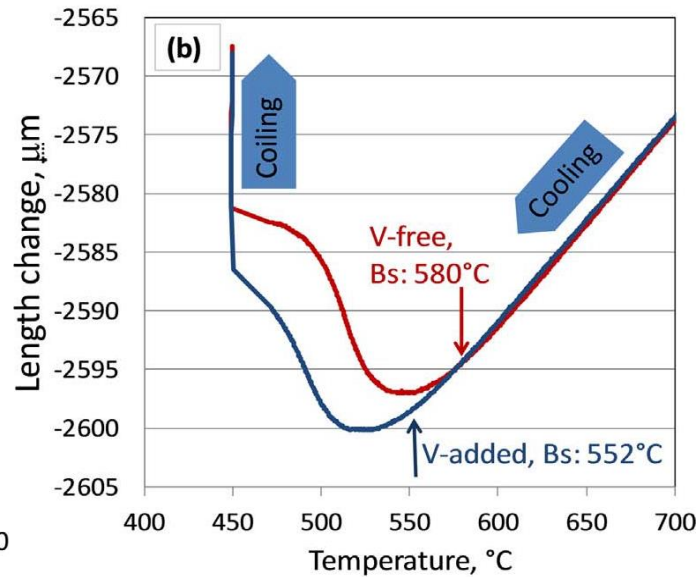
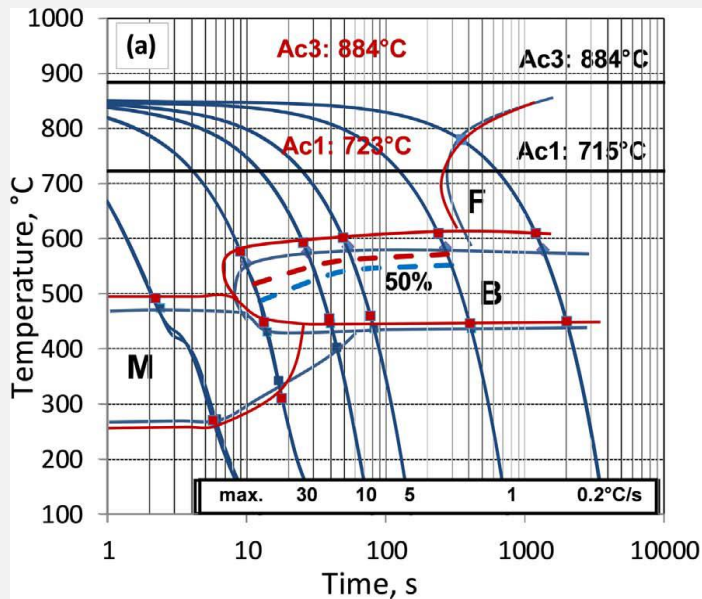
HSLA 2022, Nanjing China - 10th November 2022

Introduction - Carbide Free Bainitic Steels

- Multiphase microstructure
 - A ferritic matrix (B) with a lath morphology
 - Retained austenite (A) (interlath and granular) → stabilized at RT by carbon partitioning during the bainitic transformation (carbide precipitation delayed by Si additions)
 - Fresh martensite (M) → obtained during final cooling



Introduction - Vanadium in Very Low Carbon Bainitic Steels



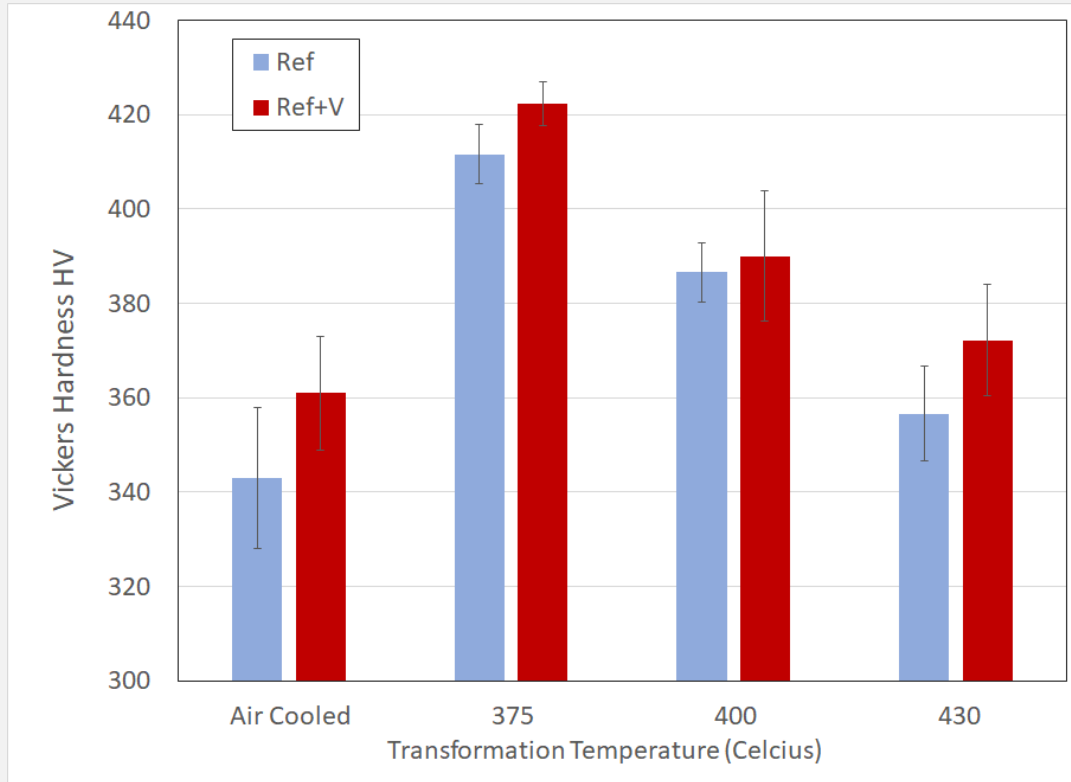
Blue lines Fe-0.06C-1.85Mn-0.16Cr-0V

Red lines Fe-0.06C-1.85Mn-0.16Cr-0.06V

Vanadium does not usually precipitate in a standard bainitic cycle, however single phase V-added low carbon bainite alloys are strengthened due to the increased hardenability and temper resistance of V in solution.

F. Fazeli, B.S. Amirkhiz, C. Scott, M. Arafin, L. Collins, *Mater. Sci. Eng. A* 720 (2018) 248–256.

Vanadium in Carbide Free Bainitic Steels



Vanadium will not precipitate in a standard CFB cycle either, however V-added CFB steels are usually stronger – why?

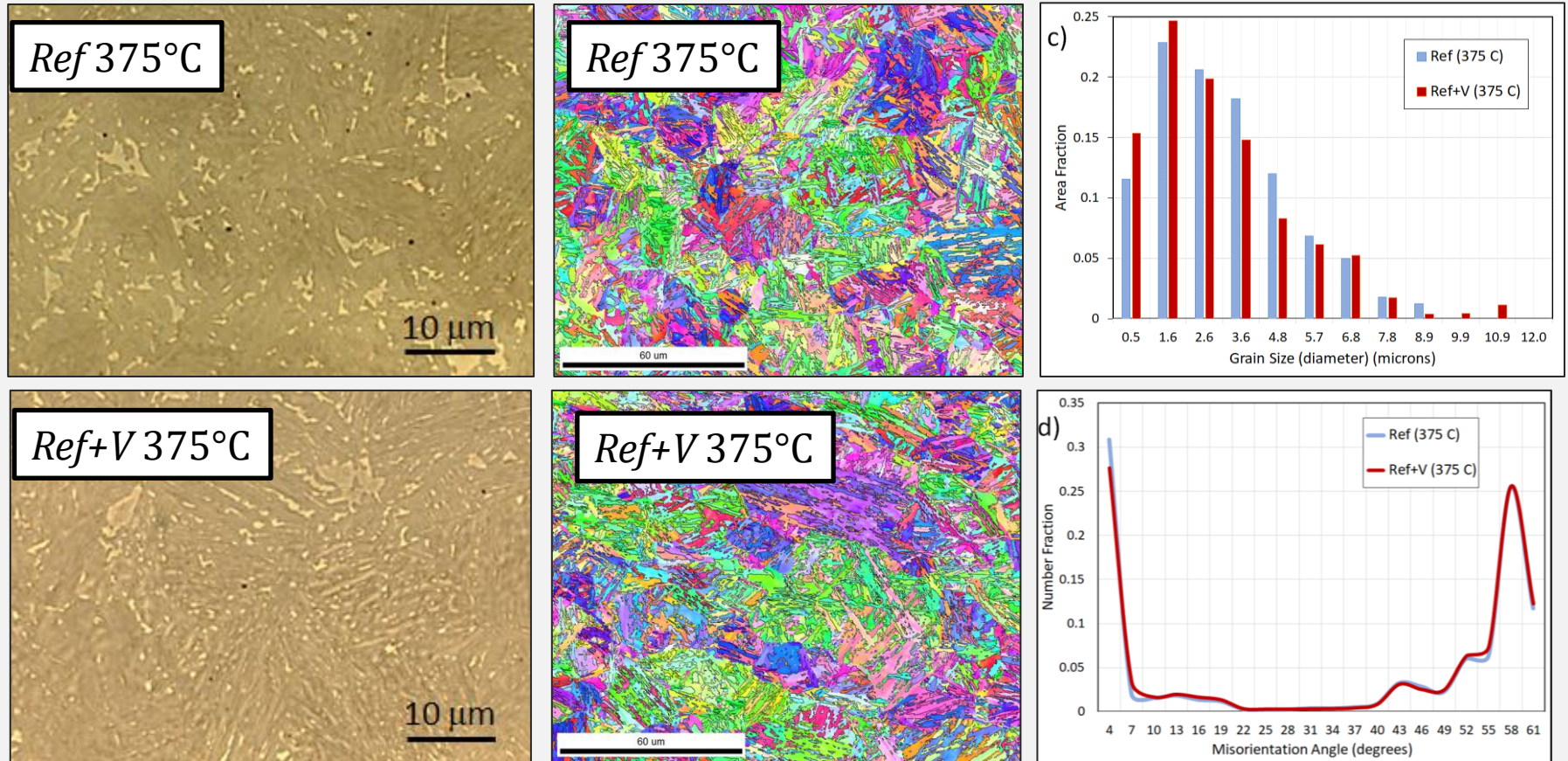
We studied the carbon distribution in bainite, austenite and martensite during and after the bainitic transformation using *in situ* High Energy Synchrotron XRD in transmission mode (mm scale) and *ex situ* using TEM-EELS (μm scale) and 3D Atom Probe (nm scale).

This is the first time these three techniques have been applied simultaneously.

Sample	C	Mn	Si	Mo	V	Al	N
Ref	0.22	2.2	1.8	0.2	-	0.01	0.0028
Ref+V	0.22	2.2	1.8	0.2	0.15	0.01	0.0027

Alloy	Ae ₁ (calc)	Ac ₁ (exp)	Ae ₃ (calc)	Ac ₃ (exp)	T _{sol} V(CN)	B _s (calc)	M _s (calc)
Ref	707°C	700°C	831°C	933°C	-	556°C	375°C
Ref+V	713°C	748°C	840°C	975°C	1067°C	556°C	375°C

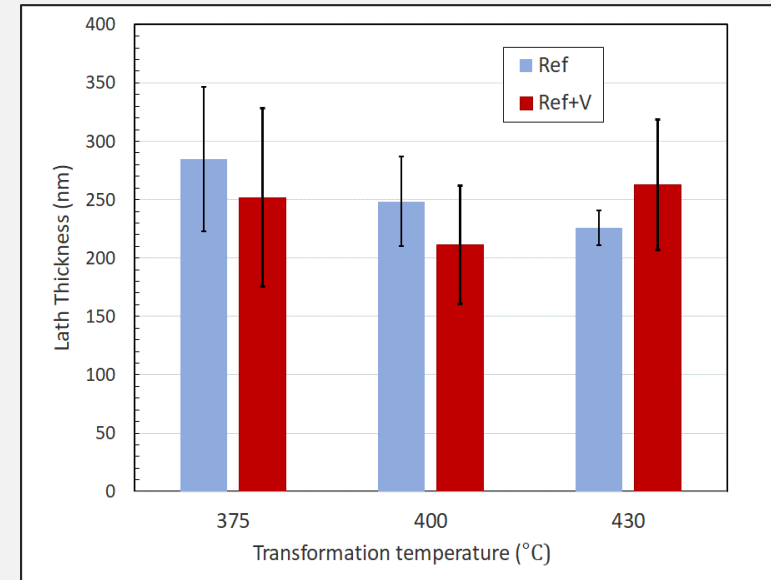
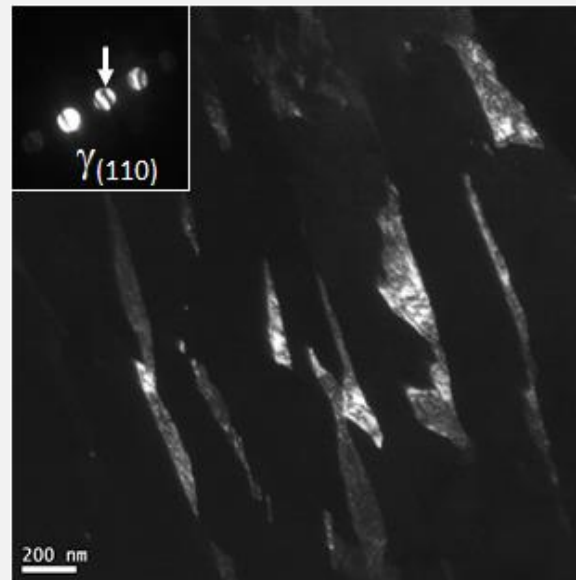
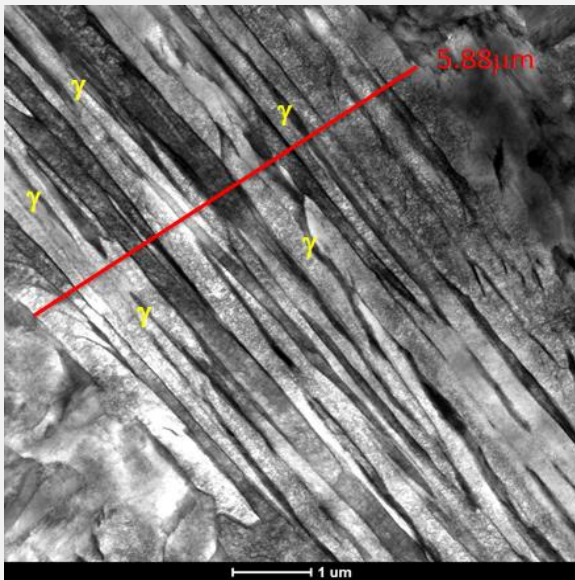
Results – SEM microstructure investigations



- Lower bainitic structures.
- No major differences between *Ref* and *Ref+V* transformed at 375°C and at 400°C except for a refinement of MA islands.

I. Pushkareva, B. Shalchi-Amirkhiz, S.Y.P. Allain, G. Geandier, F. Fazeli, M. Sztanko, C. Scott. (2020) Metals 10(3) 392.

Results – TEM microstructure investigations

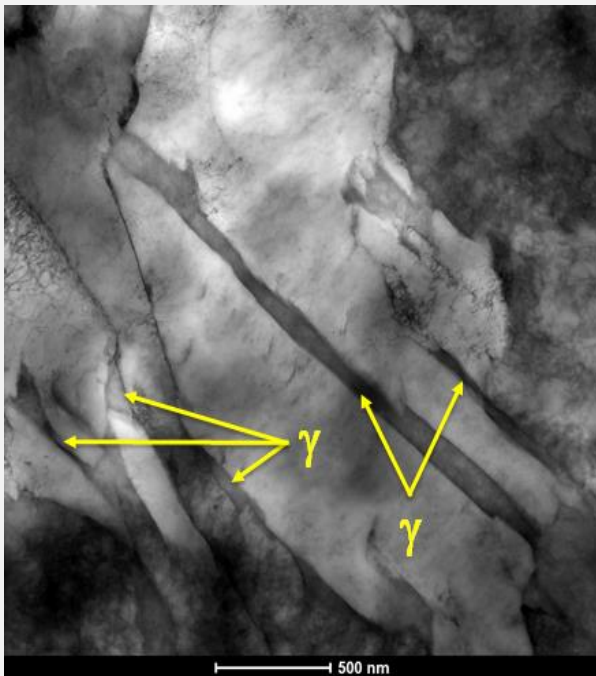


Bright field STEM image of the Ref alloy transformed at 430°C.

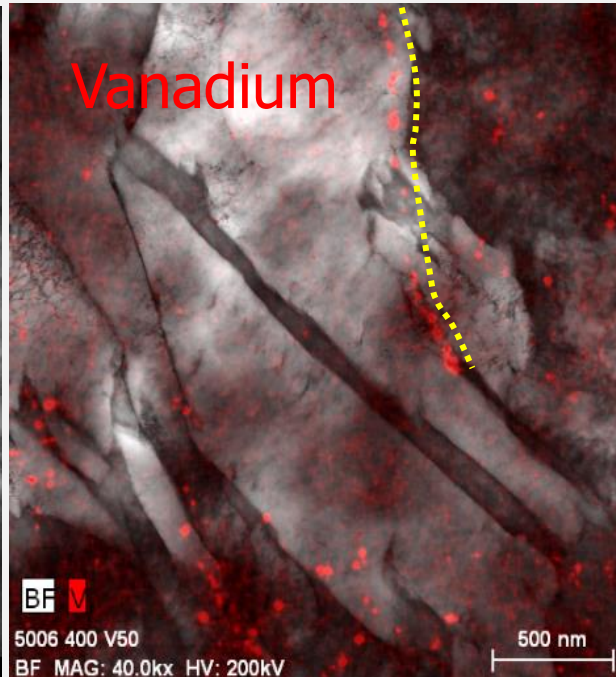
Dark field STEM image showing austenite films ($\gamma_{(110)}$ reflection) from Ref+V transformed at 430°C.

TEM shows no strong influence of Vanadium or transformation temperature on bainite lath widths

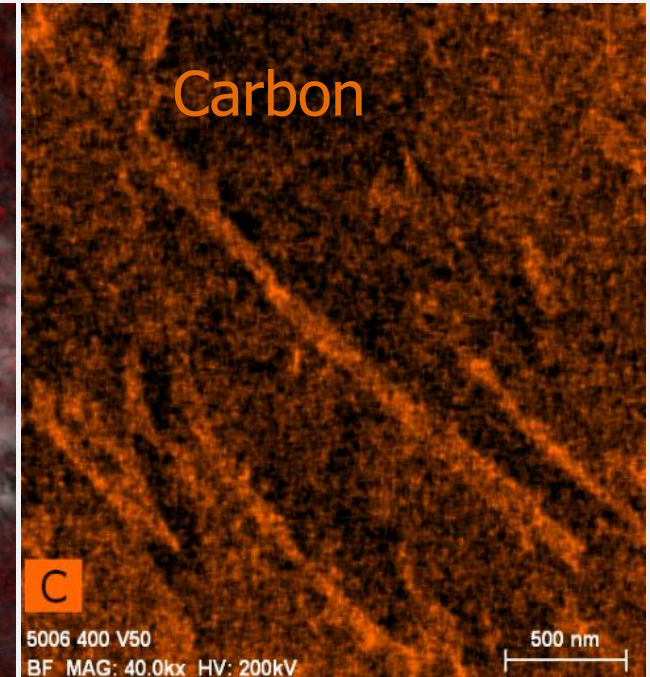
Results – TEM precipitation studies



Bright field STEM image of Ref+V transformed at 400°C.



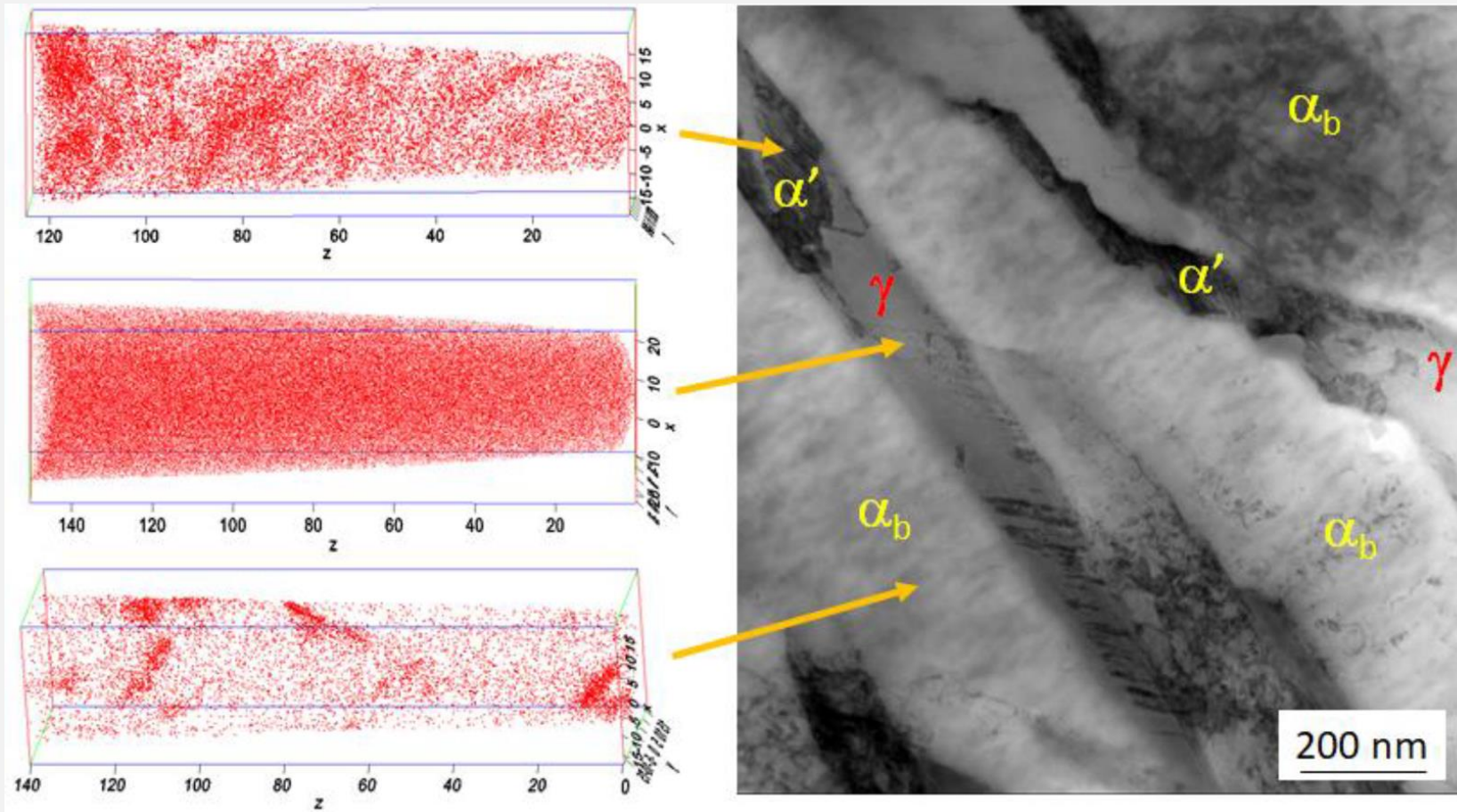
Superimposed EDX vanadium map (red) from the same area.



EDX carbon map (qualitative) from the same area.

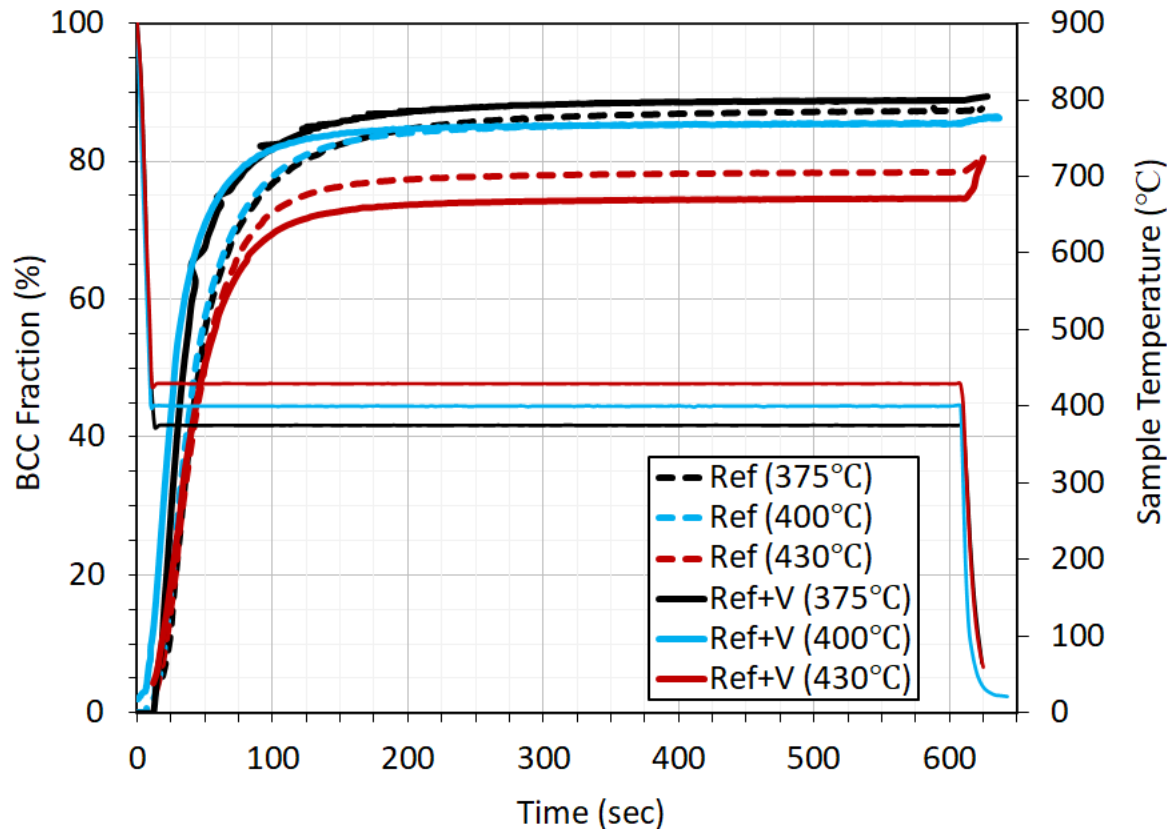
TEM shows some coarse undissolved Vanadium containing particles from the hot band – no correlation with bainite or RA. Do not appear to be VC. Very few fine precipitates were detected.

Results – Atom Probe Studies



Ref alloy transformed at 430°C. APT reconstructions show typical carbon distributions for martensite, austenite and bainitic ferrite.
Successful runs on 42 tips.

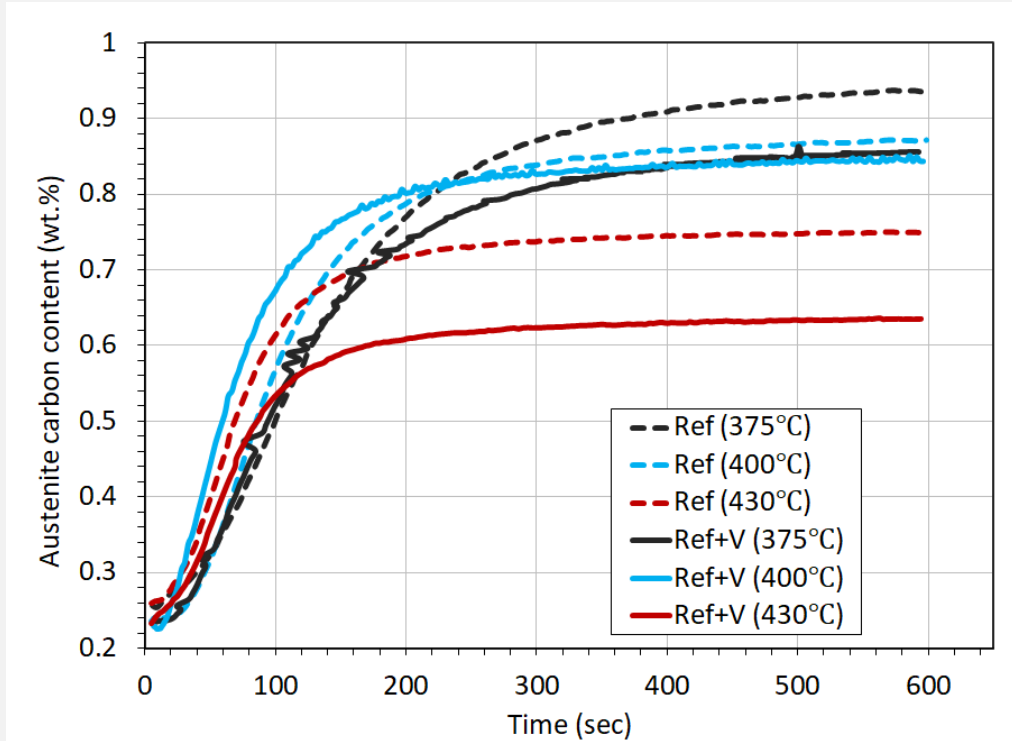
Results - HEXRD transformation kinetics and phase fractions



	T (°C)	F _b (%)	F _γ (%)	F _{α'} (%)
<i>Ref</i>	375	87.4	12.3	0.3
	400	85.6	13.5	0.9
	430	78.4	19.1	2.5
<i>Ref+V</i>	375	88.7	10.7	0.6
	400	85.3	13.8	0.9
	430	74.7	19.4	5.9

- RA fraction increases as transformation temperature decreases
- Sigmoidal kinetics and temperature dependent stasis (no significant effect of V)
- Final martensite transformations occur at 430°C only (V slightly increases F_{α'})

Results - HEXRD carbon content at stasis in austenite



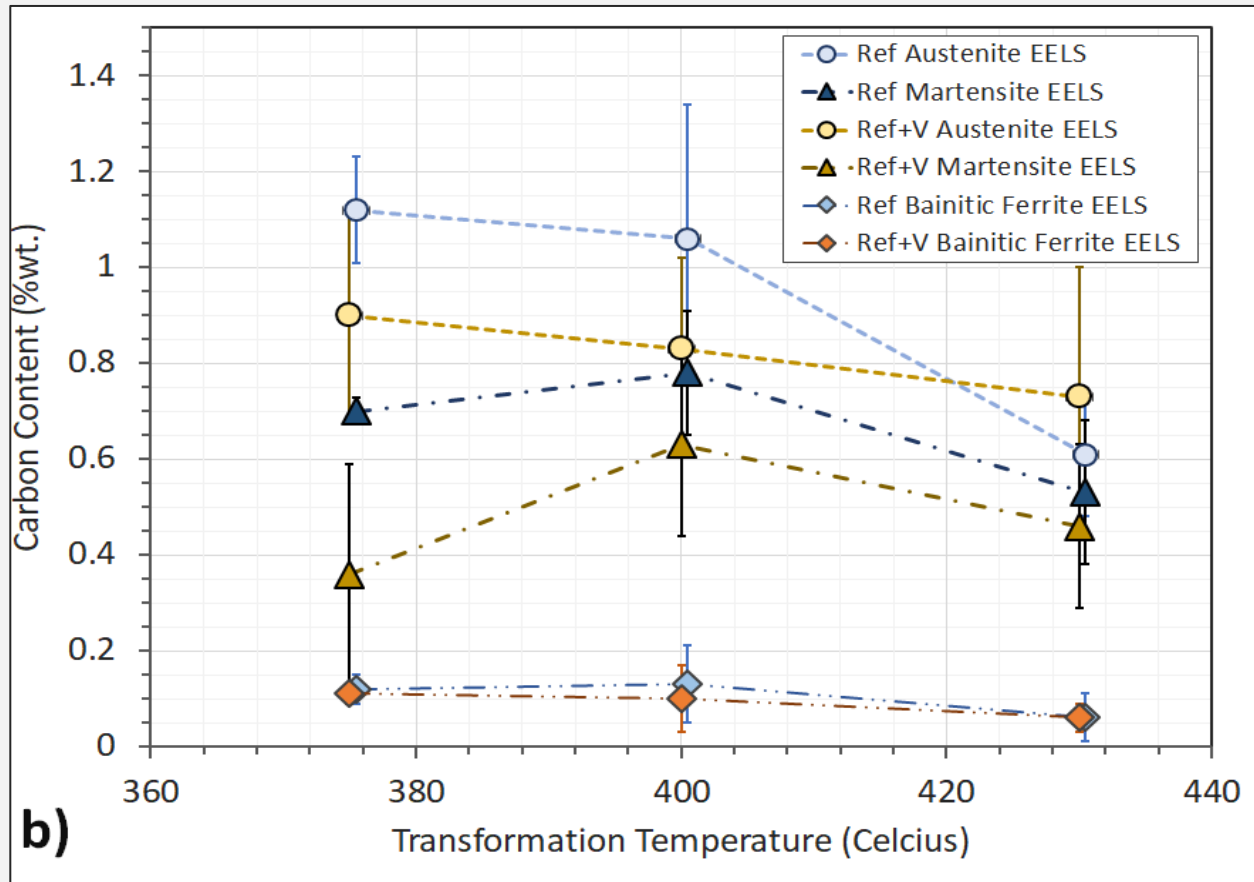
	T (°C)	F _γ HEXRD @ 600s (%)	C _γ HEXRD (wt.%)	C _b HEXRD (wt.%)
Ref	375	12.6	0.94	0.12
	400	14.4	0.87	0.11
	430	21.6	0.75	0.07
Ref+V	375	11.3	0.86	0.14
	400	14.7	0.85	0.11
	430	25.3	0.63	0.08

$$C_{\gamma} = C_0 + \frac{a_{\gamma} - a_{\gamma 0}(T)}{0.033}$$

- Carbon content in austenite determined considering the relative evolution of the lattice parameter
- Note that carbon partitioning to austenite continues long after the transformation stops. **V addition appears to slow down the partitioning kinetics.**
- No ferrite is formed during cooling so the carbon content in bainitic ferrite at stasis (375-430°C) can be obtained from a simple mass balance equation:

$$C_0 = C_b(1 - F_{\gamma}) + C_{\gamma} F_{\gamma}$$

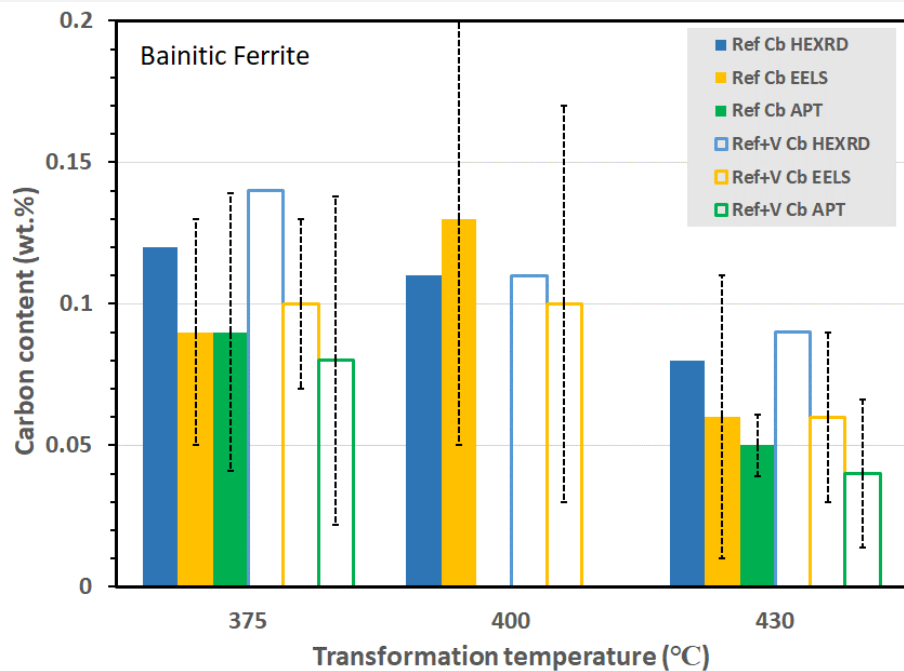
Results - TEM EELS Carbon Measurements



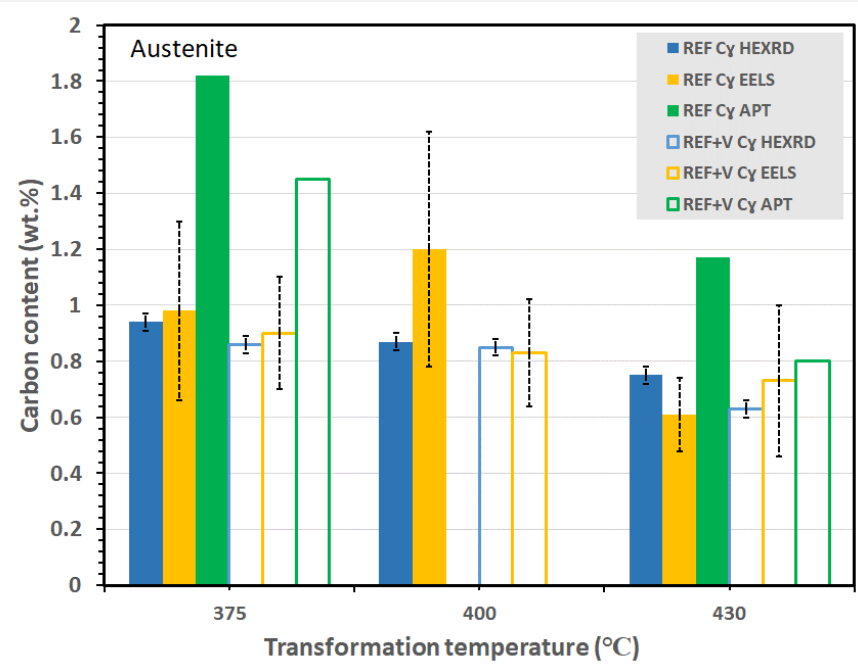
- EELS measurement precision is +/- 4% and the detection limit for solute carbon is 0.02-0.03 wt. %
- Large scattering on C_{γ} measurements reflects a real distribution. Mean values slightly higher than obtained by HEXRD
- Carbon in martensite is lower (less enriched austenite transformed during final cooling)

Results – HEXRD v EELS v APT

Alloy	Transformation Temperature	C _γ HEXRD (wt.%)	C _γ EELS (wt.%)	C _γ APT (wt.%)	C _α EELS (wt.%)	C _b HEXRD (wt.%)	C _b EELS (wt.%)	C _b APT (wt.%)
Ref	375°C	0.94	0.98 (9)	1.82 (2)	0.70 (1)	0.12	0.09 (12)	0.11 (12)
Ref	400°C	0.87	1.20 (7)	-	0.78 (4)	0.11	0.13 (4)	-
Ref	430°C	0.75	0.69 (8)	1.17 (1)	0.49 (5)	0.08	0.07 (10)	0.05 (7)
Ref+V	375°C	0.86	0.90 (9)	1.45 (2)	0.36 (3)	0.14	0.10 (13)	0.08 (12)
Ref+V	400°C	0.85	0.83 (10)	-	0.63 (9)	0.11	0.10 (3)	-
Ref+V	430°C	0.63	0.73 (8)	0.80 (1)	0.46 (7)	0.09	0.06 (5)	0.04 (9)

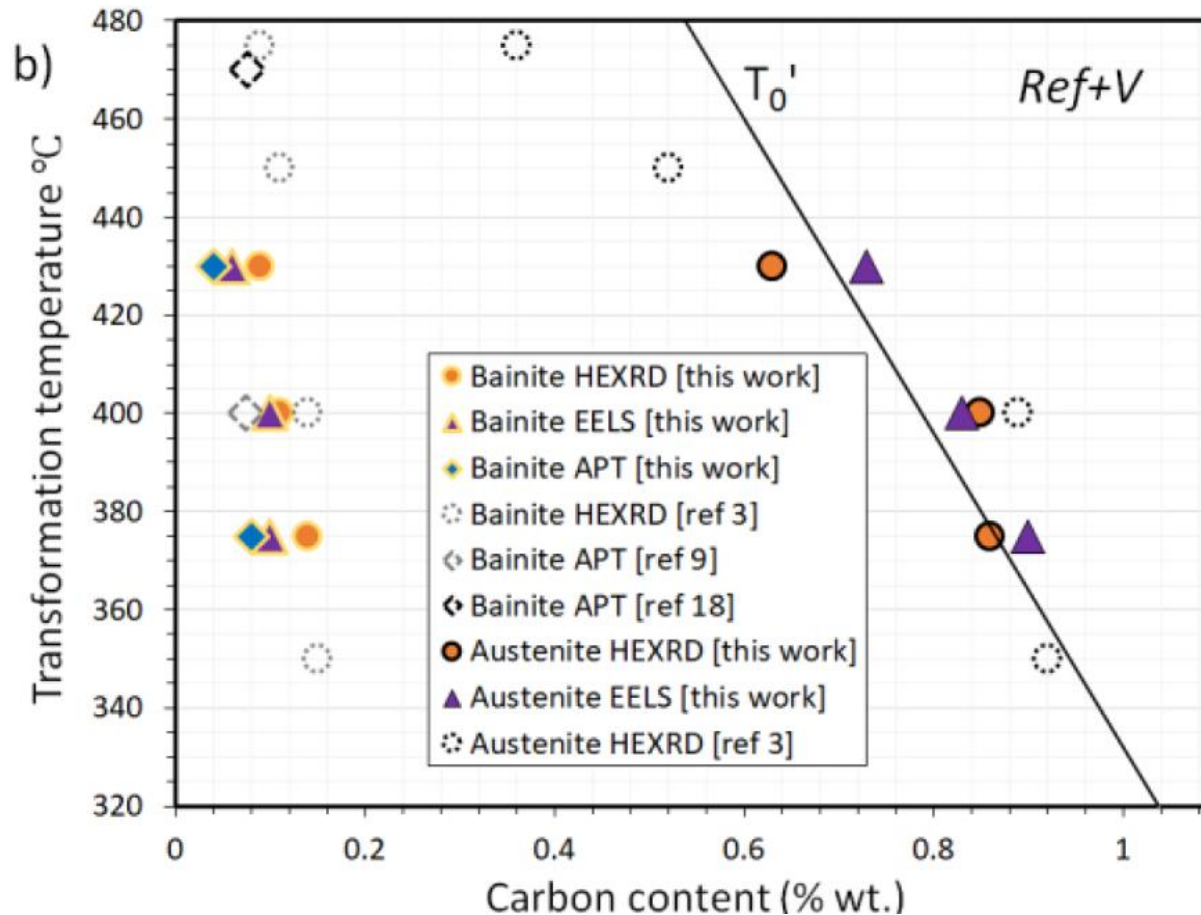


No clear influence of V on C_b



Adding V *slightly* decreases C_γ

Mean values – comparison with literature



[3] R. Rementeria, J.D. Poplawsky, M.M. Aranda, W. Guo, J.A. Jimenez, C. Garcia-Mateo, F.G. Caballero, *Acta Mater.* 125 (2017) 359–368.

[9] E.V. Pereloma, *Mater. Sci. Technol. (United Kingdom)* 32 (2016) 99–103.

[18] G. Miyamoto, K. Shinbo, T. Furuhashi, *Scr. Mater.* 67 (2012) 999–1002.

- Consistent results, HEXRD appears give higher C_b values. This is not due to lattice tetragonality.
- High fraction of carbon trapped in bainite (even with low C_0), much higher than PE solubility in bcc → this C is lost for austenite stabilization although it contributes to bainitic ferrite strength.

Structure-properties modelling

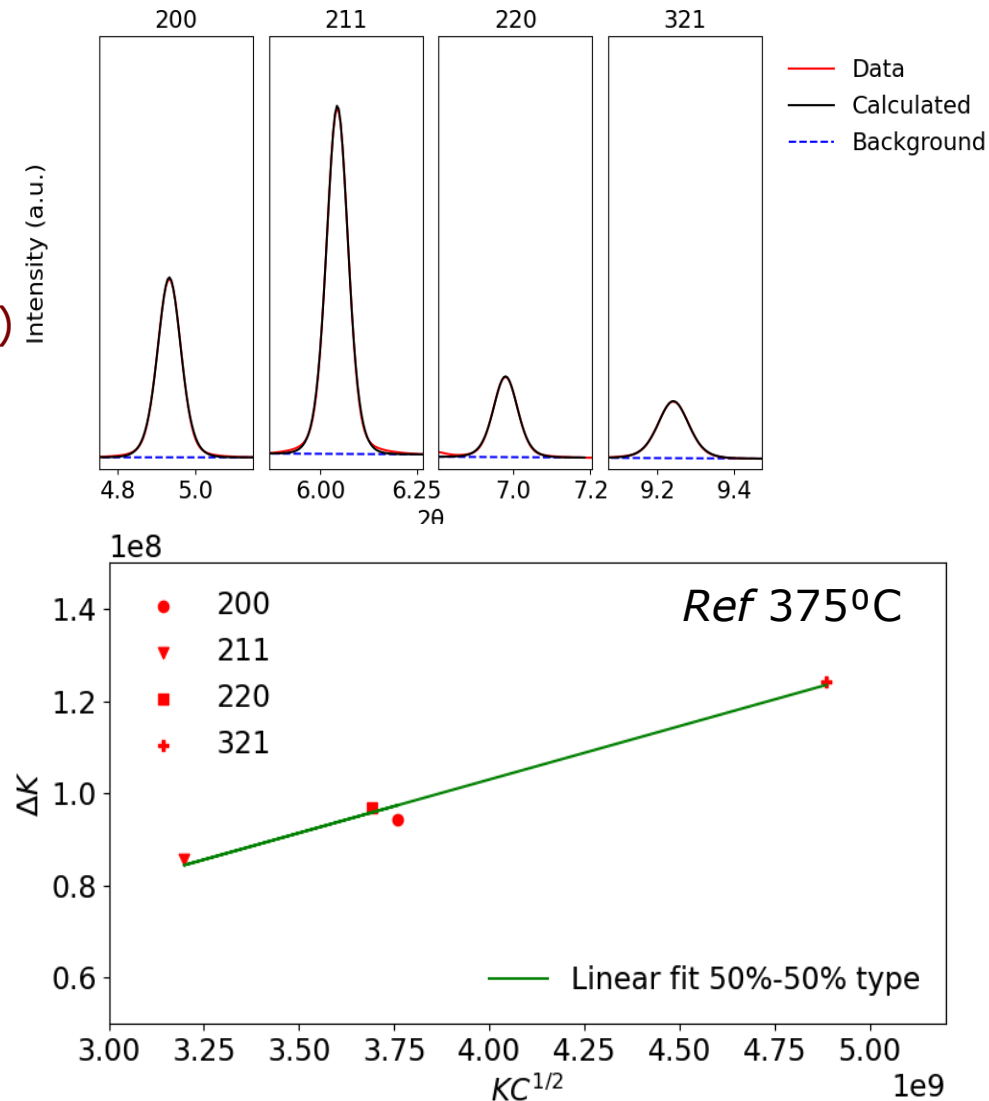
Vanadium effect on CFB microstructure:

- Does not change the transformation kinetics
- Does not change the final fraction of RA
- RA carbon content is slightly lower
- Fractionally more martensite is formed (at 430°C)
- No change in bainitic lath widths
- No measurable change in bainitic ferrite carbon content
- No significant precipitation of V

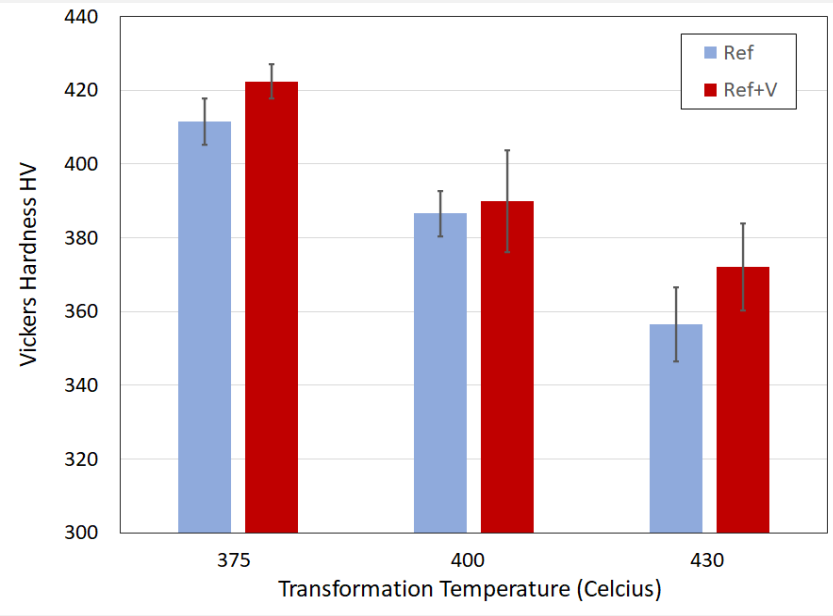
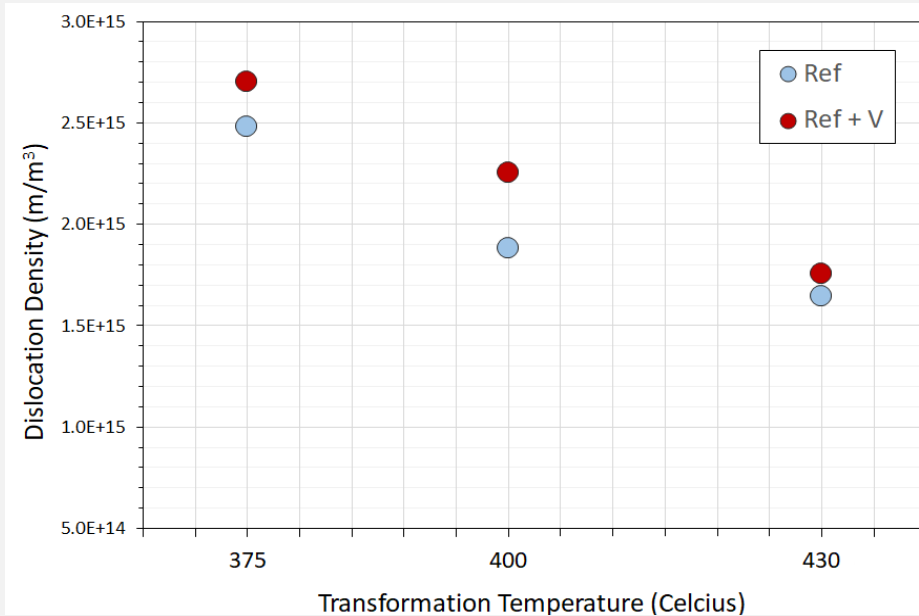
Only dislocation density left to check

Use modified Williamson-Hall method:

*J. Macchi, G. Geandier, J. Teixeira, S. Denis, F. Bonnet, S. Y. P. Allain
Materialia Volume 26, December 2022, 101577.*



Structure-properties modelling

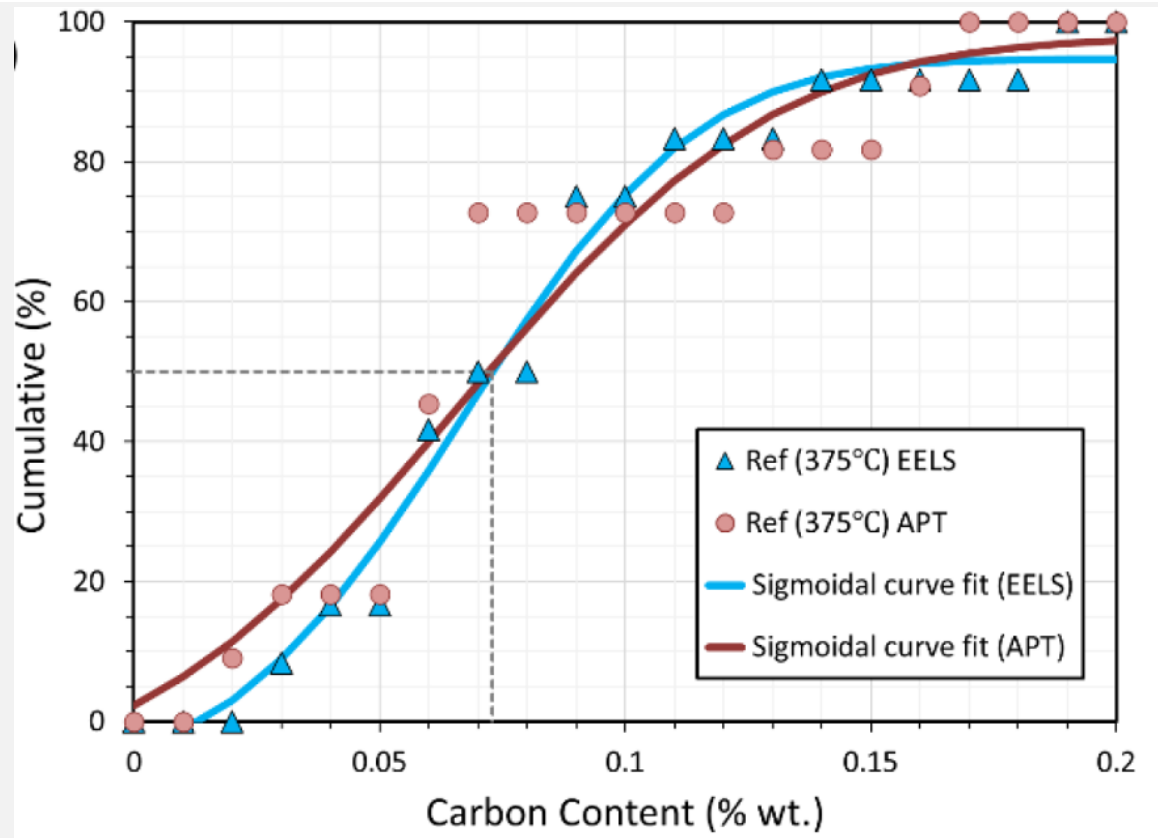


- Dislocation density measurements are strongly correlated to the measured hardness behaviour.
- Inverse correlation of hardness with RA mean carbon content.
- Vanadium additions increase the value of ρ at all transformation temperatures tested.
- Similar to the effect in very low C bainite – but the mechanism(s) is/are quite different!

At least 3 competing mechanisms: V slows down C partitioning to austenite. RA is softer, easier to deform during the transformation (reduces ρ) but less stable so more martensite forms during cooling (increases ρ , especially at high T). However, solute V ($+\Delta C_b$?) slows down static + dynamic recovery in bainite (increases ρ).

Going further – statistical analysis

Alloy	Transformation Temperature	C_γ HEXRD (wt.%)	C_γ EELS (wt.%)	C_γ APT (wt.%)	C_α EELS (wt.%)	C_b HEXRD (wt.%)	C_b EELS (wt.%)	C_b APT (wt.%)
Ref	375°C	0.94	0.98 (9)	1.82 (2)	0.70 (1)	0.12	0.09 (12)	0.11 (12)



- The concentration of excess carbon in bainitic ferrite varies widely from region to region.
- Excess carbon trapped in bainite does not follow a normal distribution – bias is towards lower carbon contents.
- Could this be correlated with bainite lath width ?

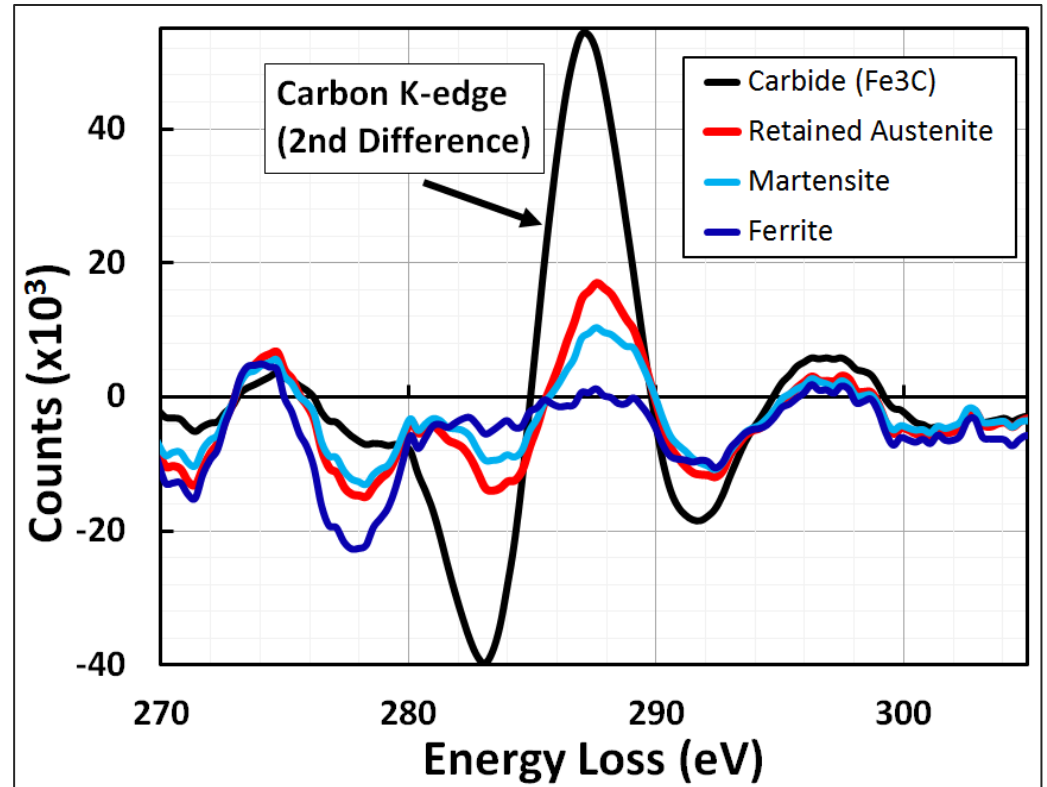
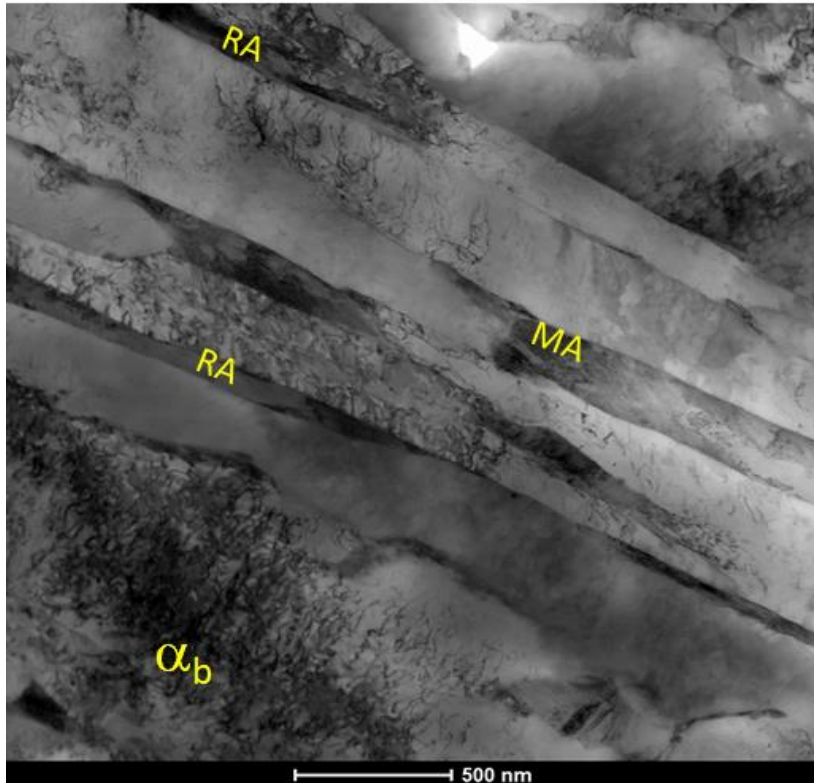
Conclusions

- The influence of V on the microstructure evolution of isothermally transformed CFB steels has been studied.
- Three advanced techniques at different length scales have been coupled to investigate the carbon content in the constituent phases (bainitic ferrite / austenite / martensite) of CFB steels.
- The three methods give similar results showing that a large fraction of carbon C_b remains trapped in bainitic ferrite even at high transformation temperatures. C_b is not strongly influenced by vanadium additions.
- Effect of vanadium in CFB (solid solution):
 - No significant effect on kinetics, stasis, final phase fractions bainite lath size or C_b
 - No evidence of precipitation strengthening
 - Slows the rate of carbon partitioning to austenite, slight reduction in final C_γ
 - Increases dislocation density ρ in BF at all tested temperatures
 - Final mechanical properties correlate well with ρ (and inverse correlation with C_γ)

Thank you for your attention!

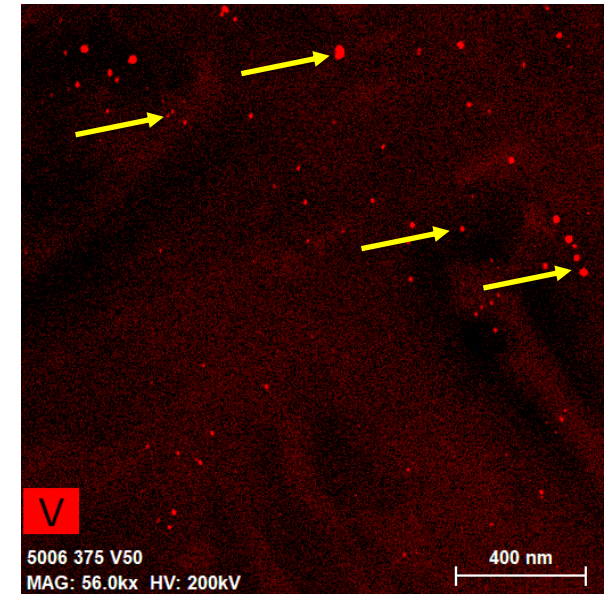
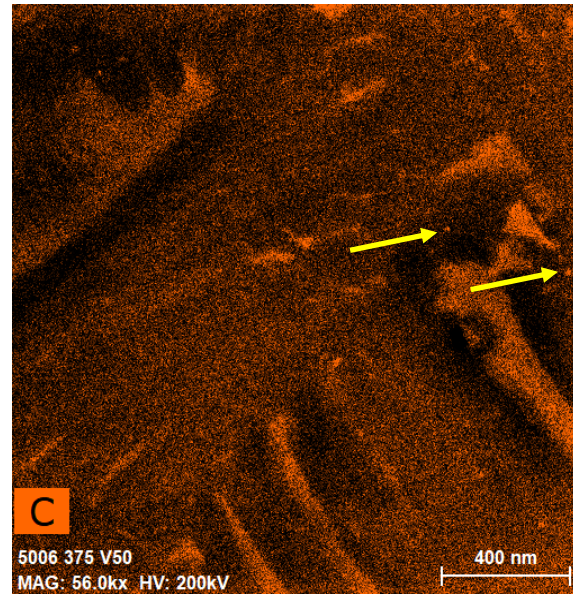
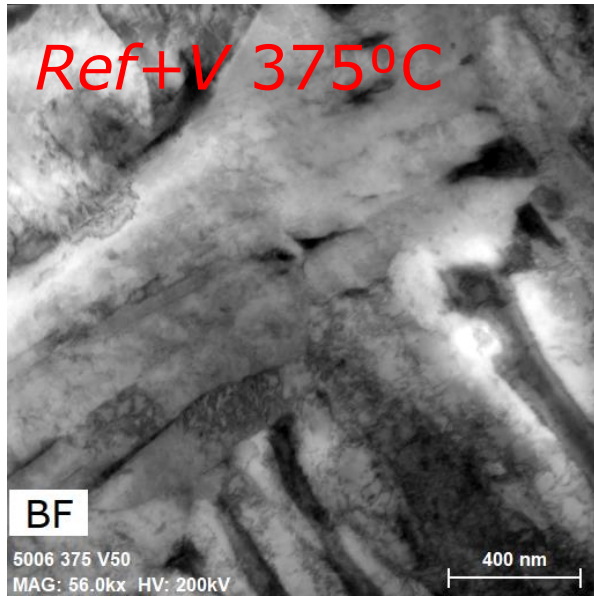
Backup slides

Materials and Methods – TEM EELS

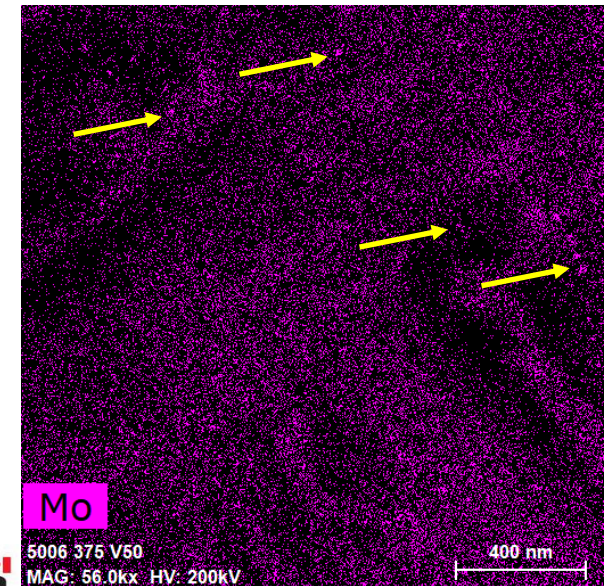


- TEM investigations (CanmetMATERIALS)
 - Tecnai Osiris 200 keV X-FEG (S)TEM / FEI FS1 EELS analysis system
 - EELS using a standards based second difference acquisition technique developed by the authors

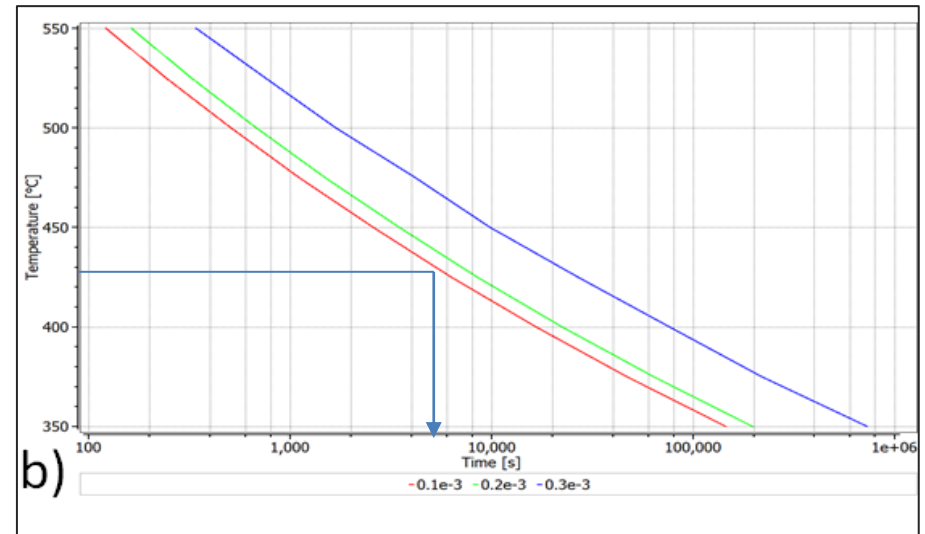
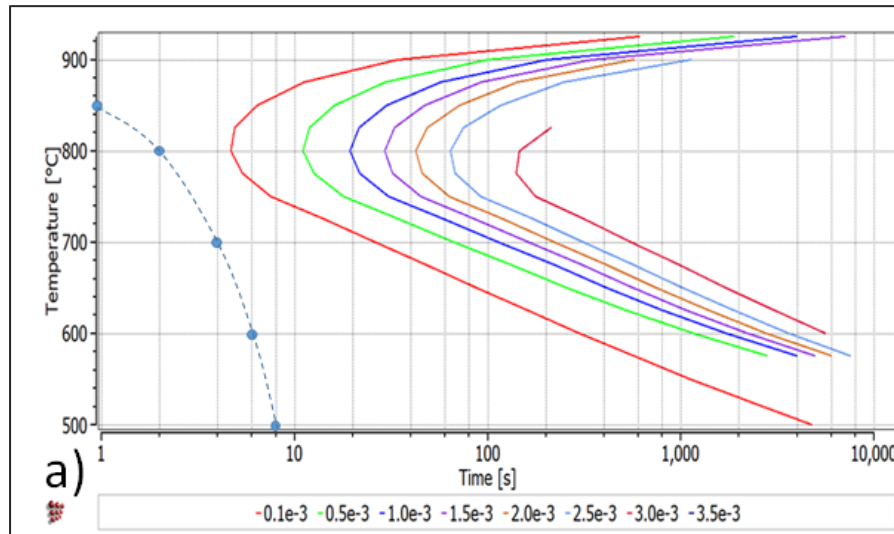
Results – TEM precipitation studies



Most of the particles observed in *Ref+V* specimens transformed at 400°C and 430°C contained vanadium with small amounts of carbon and nitrogen being occasionally detected. However, at 375°C some of the particles contained V+Mo. Mo enrichment of V(C,N) is only significant at temperatures below 800°C, suggesting that these particles formed in ferrite



Results – kinetic precipitation modelling



- a) $V(C,N)$ PTT in *Ref+V* ferrite containing $1 \times 10^{14}/\text{m}^2$ dislocations.
- b) Low temperature PTT for $V(C,N)$ in *Ref+V* bainitic ferrite containing $5 \times 10^{14}/\text{m}^2$ dislocations

Kinetic modelling predicts that no $V(C,N)$ should form for transformation times $< 5000\text{s}$